

Uses of PFAS and their alternatives: applications of essentiality and functional substitution

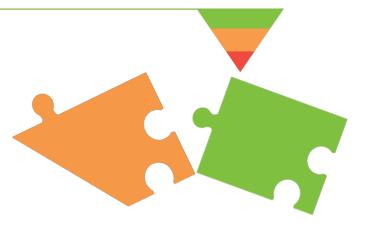


## **Ian Cousins**

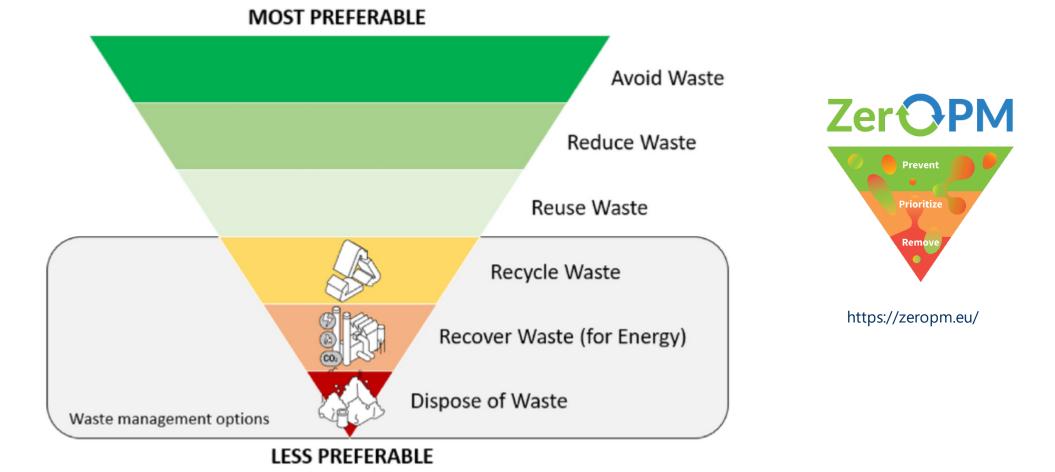
Stockholm University, Sweden **REMTEC & EMERGING CONTAMINANTS SUMMIT**, Westminster, Colorado, US, 15<sup>th</sup> October 2024



This project has received funding from the European Union's Horizon 2020 research and innovation programme under grant agreement No 101036756.



## Waste hierarchy

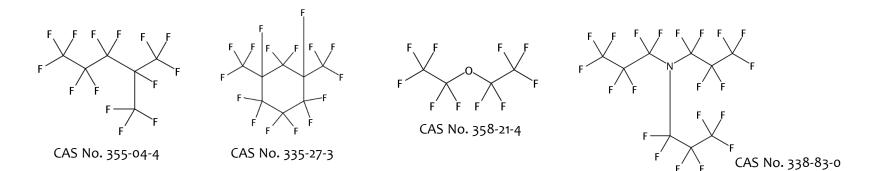


## What are PFAS?

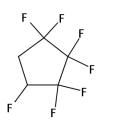
- Buck et al. (2011) first definition
- OECD (2021): broader definition
  - "...the fluorinated substances that contain at least one fully fluorinated methyl or methylene carbon atom..." i.e. substances are PFAS that have **at least one -CF<sub>2</sub>- or -CF<sub>3</sub>** moiety in their structure
  - used as regulatory definition in the EU universal restriction proposal, but if PFAS are mineralizable then they are exempted
- US EPA (2023) definition less broad; two connected fluorinated carbons
- Many thousands of **structurally diverse PFAS** in use in society
  - polymers & non-polymers; neutral, anionic, cationic & zwitterionic; solids, liquids & gases; reactive & inert; soluble & insoluble; volatile & involatile; mobile & immobile; bioaccumulative & non-bioaccumulative; highly toxic and relatively non-toxic

## **Volatile neutral PFAS**

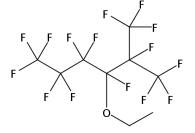
Perfluoroalkanes (PFCs), perfluoroethers and perfluoroalkylamines



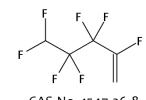
Certain hydrofluoro-carbons (HFCs), -ethers (HFEs) and -olefins (HFOs)



CAS No. 15290-77-4



CAS No. 297730-93-9

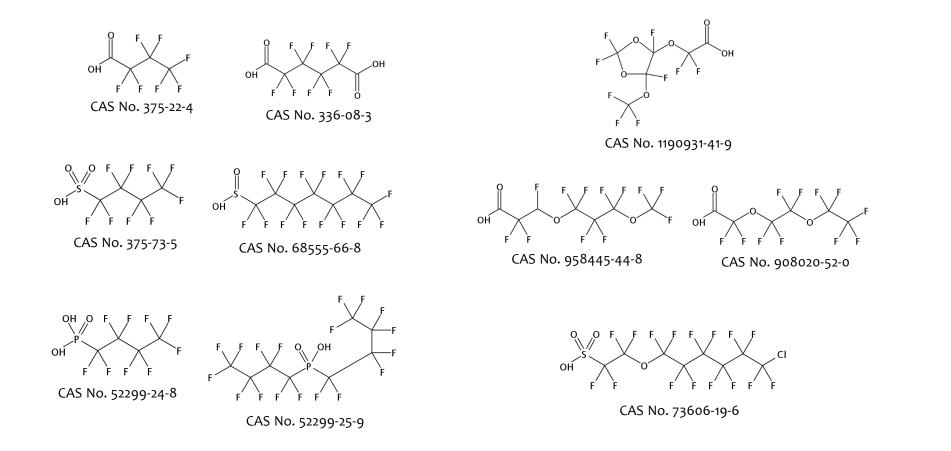


CAS No. 1547-26-8

## **Involatile anionic PFAS**

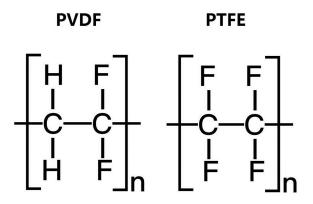
• Per- and polyfluoroalkyl acids (PFAAs)

Per- and polyfluoroalkylether acids

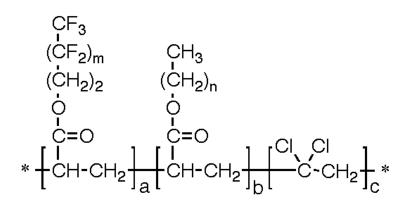


## **Fluorinated polymers**

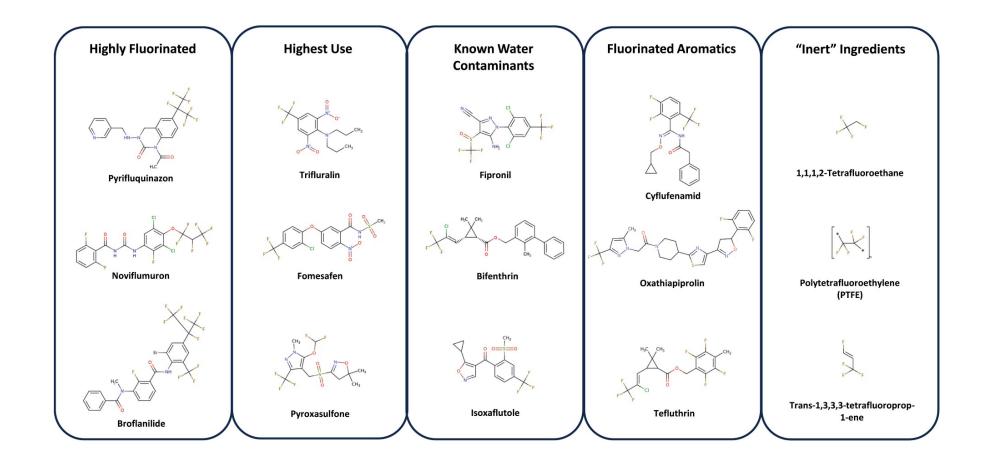
- Fluoropolymers (e.g. PTFE) (F in the backbone)
  - high molecular weight, stable, inert, insoluble, involatile, immobile, do not cross biological membranes, low leachables
- Side-chain fluorinated polymers (as used in textiles, carpets, food packaging)
  - non-fluorinated co-polymer backbone with fluorinated side chains
  - leachable PFAS, stable?



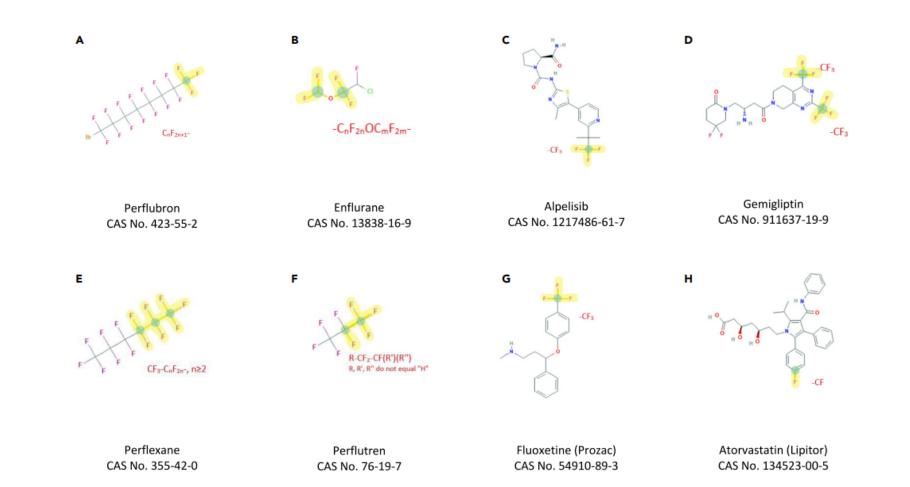
Polyvinylidene fluoride Polytetrafluoroethylene



## And many more: pesticides



## And many more: pharmaceuticals



## Attempt to make comprehensive overview by OECD

		رـــ perfluoroalkyl carboxylic acids (PFCAs), ۲٫۹۶٫۹٫۰٬۰۲۵	Nomanelatura including acconume well encored by Busk at al. (accu)			
			Nomenclature including acronyms well covered by Buck et al. (2011)			
	perfluoroalkyl acids (PFAAs; including perfluoroalkylether – acids, PFEAAs)	- perfluoroalkane sulfonic acids (PFSAs), C <sub>n</sub> F <sub>2n+1</sub> -SO <sub>3</sub> H	Nomenclature including acronyms partially covered by Buck et al. (2011)			
		- perfluoroalkyl phosphonic acids (PFPAs), C <sub>n</sub> F <sub>2n+1</sub> -PO <sub>3</sub> H <sub>2</sub>	Common nomenclature including acronyms exists  Indicating that it is a synthesis pathway, instead of examples Text Polymeric PFASs			
		– perfluoroalkyl phosphinic acids (PFPIAs), (C <sub>n</sub> F <sub>2n+1</sub> )(C <sub>m</sub> F <sub>2m+1</sub> )-PO <sub>2</sub> H				
ſ		eg. C <sub>2</sub> F <sub>5</sub> OC <sub>2</sub> F <sub>4</sub> OCF <sub>2</sub> COOH	Note: This figure is intended to be comprehensive, but not exhaustive; in other words, there are			
		perfluoroalkylether sulfonic acids (PFESAs), e.g. C <sub>6</sub> F <sub>13</sub> OCF <sub>2</sub> CF <sub>2</sub> SO <sub>3</sub> H	other groups of PFASs that are not captured in this figure.			
		— perfluoroalkyl dicarboxylic acids (PFdiCAs), HOOC-C <sub>n</sub> F <sub>2n</sub> -COOH	It is recommended that polyfluoroalkyl acids use the acronym PolyFAA to better distinguish from perfluoroalkyl acids. Depending on the fluorination degree and location, polyfluoroalkyl carboxylic			
		— perfluoroalkane disulfonic acids (PFdiSAs), HO <sub>3</sub> S-C <sub>n</sub> F <sub>2n</sub> -SO <sub>3</sub> H	acids may either be highly persistent themselves, or act as precursors to other PFAAs such as			
		perfluoroalkane sulfinic acids (PFSIAs), C <sub>n</sub> F <sub>2n+1</sub> -SO <sub>2</sub> H	PFdiCAs, under natural conditions. ** Strictly speaking, these substances are not fluorotelomers, as they are not derived from the			
	polyfluoroalkyl acids (PolyFAAs; including polyfluoroalkylether acids, PolyFEAAs)*	polyfluoroalkyl carboxylic acids (PFCAs or PolyFCAs), e.g. H-C <sub>n</sub> F <sub>2n</sub> -COOH, n>1 polyfluoroalkylether carboxylic acids (PFECAs), e.g. CF <sub>3</sub> OC <sub>3</sub> F <sub>6</sub> OCHFCF <sub>2</sub> COOH polyfluoroalkylether sulfonic acids (PFESAs), e.g. CIC <sub>6</sub> F <sub>12</sub> OCF <sub>2</sub> CF <sub>2</sub> SO <sub>3</sub> H	telomerization process. Despite this, they are termed here "n:1 fluorotelomer-based" substances for readability. Future work may consider to identify more proper terminology for this group of PFASs. *** Depending on the type of linkages between fluorinated side chain(s) and aromatic ring(s), some side-chain fluorinated aromatics may act as precursors to PFAAs or PFEAAs. **** Depending on the molecule structure, one may belong to PFAAs, PolyFAAs, PFAA precursors, or other groups that are not described here.			
		- n:1 fluorotelomer alcohols, C <sub>n</sub> F <sub>2n+1</sub> CH <sub>2</sub> OH**				
		- perfluoroalkanoyl fluorides (PACFs), C <sub>n</sub> F <sub>2n+1</sub> COF	PACF-based substances, C <sub>n</sub> F <sub>2n+1</sub> CO <sub>2</sub> -R e.g. (meth)acrylate, urethane or oxetane polymers, silicones			
PFASs —		– perfluoroalkyl iodides (PFAIs), C <sub>n</sub> F <sub>2n+1</sub>	n:2 fluorotelomer-based substances, C <sub>n</sub> F <sub>2n+1</sub> CH <sub>2</sub> CH <sub>2</sub> -R non-polymers			
		– perfluoroalkane sulfonyl fluorides (PASFs), C <sub>n</sub> F <sub>2n+1</sub> SO <sub>2</sub> F	$\rightarrow$ PASF-based substances, C <sub>n</sub> F <sub>2n+1</sub> SO <sub>2</sub> -R $R = NH, NHCH_2CH_2OH, etc.$			
		– perfluoroalkylether non-polymers, e.g. C <sub>4</sub> F <sub>9</sub> OC <sub>2</sub> F <sub>4</sub> OC <sub>2</sub> F <sub>4</sub> OCF <sub>2</sub> -CH <sub>2</sub> OH, CAS No. 317817-24-6				
	PFAA precursors –	- perfluoroalkylether side-chain fluorinated polymers				
		— perfluoroalkenes (C <sub>n</sub> F <sub>2n</sub> , n>2)	perfluoroalkene derivatives, e.g. ↓ [(CF <sub>3</sub> ) <sub>2</sub> CF] <sub>2</sub> C=C(CF <sub>3</sub> )OC <sub>6</sub> H <sub>4</sub> SO <sub>3</sub> Na), CAS No. 70829-87-7			
		— semifluorinated alkanes (SFAs), C <sub>n</sub> F <sub>2n+1</sub> -C <sub>m</sub> H <sub>2m+1</sub>	· 3 <sup>2</sup> · 2 · 3 <sup>2</sup> 0 4 3 ··· · · · · · ·			
		hydrofluorocarbons (HFCs, e.g. $C_nF_{2n+1}-C_mH_{2m+1}$ ), hydrofluoroethers (HFEs, e.g. $C_nF_{2n+1}-C_mH_{2m+1}$ ), hydrofluoroolefins (HFOs, e.g. $C_nF_{2n+1}-CH=CH_2$ ) that have a perfluoroalkyl chain				
		- perfluoroalkyl (e.g. C <sub>n</sub> F <sub>2n+1</sub> C(O)C <sub>m</sub> F <sub>2m+1</sub> ) and semi-fluorinated ketones (e.g. C <sub>n</sub> F <sub>2n+1</sub> C(O)C <sub>m</sub> H <sub>2m+1</sub> )				
		perfluoroalkyl alcohols (C <sub>n</sub> F <sub>2n+1</sub> OH), e.g. (CF <sub>3</sub> ) <sub>3</sub> C-OH, CAS No. 2378-02-1				
		fluoropolymers (FPs)	polytetrafluoroethylene (PTFE)			
		- perfluoropolyethers (PFPEs), e.g. HOCH <sub>2</sub> O-(C <sub>m</sub> F <sub>2m</sub> O) <sub>n</sub> -CH <sub>2</sub> OH	— polyvinylidene fluoride (PVDF)			
			— fluorinated ethylene propylene (FEP)			
	other PFASs	- perfluoroalkanes (C <sub>n</sub> F <sub>2n+2</sub> )	— perfluoroalkoxyl polymer (PFA)			
		— perfluoroalkyl- <i>tert</i> -amines (C <sub>n</sub> F <sub>2n+1</sub> ) <sub>3</sub> N	Other FPs			
		— perfluoroalkylethers (e.g. C <sub>n</sub> F <sub>2n+1</sub> OC <sub>m</sub> F <sub>2m+1</sub> )				
		others****				

...

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# So how many PFAS are there? It depends on how you count...

- >7 million on PubChem database
  - many only listed in patents
- 4830 OECD 2018 list
- ca. 14 000 in US EPA CompTox list
- > 10 000 in the EU PFAS Restriction Proposal
- **1 670** with identified uses
- **531** are REACH registered (>1 tonne/a)

## **Production volumes of PFAS**

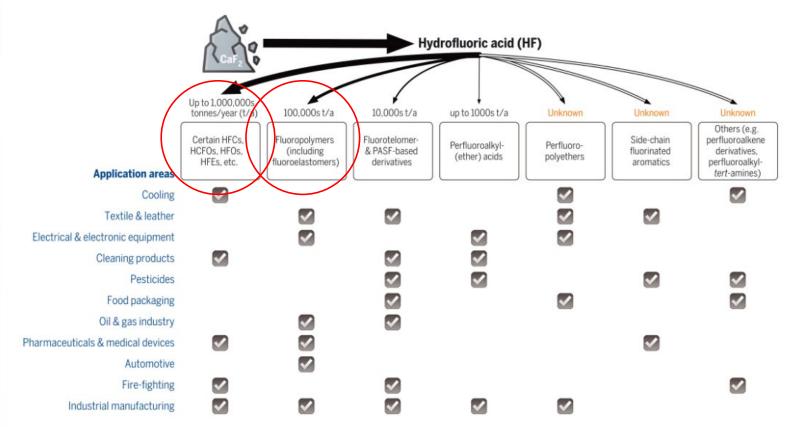


Fig. 1. Non-exhaustive summary of PFAS manufacturing, from production to consumer use.

- Industry wants most to protect uses of fluoropolymers and fluorinated gases
- Definition of PFAS changes to exclude them from PFAS definition in some US States

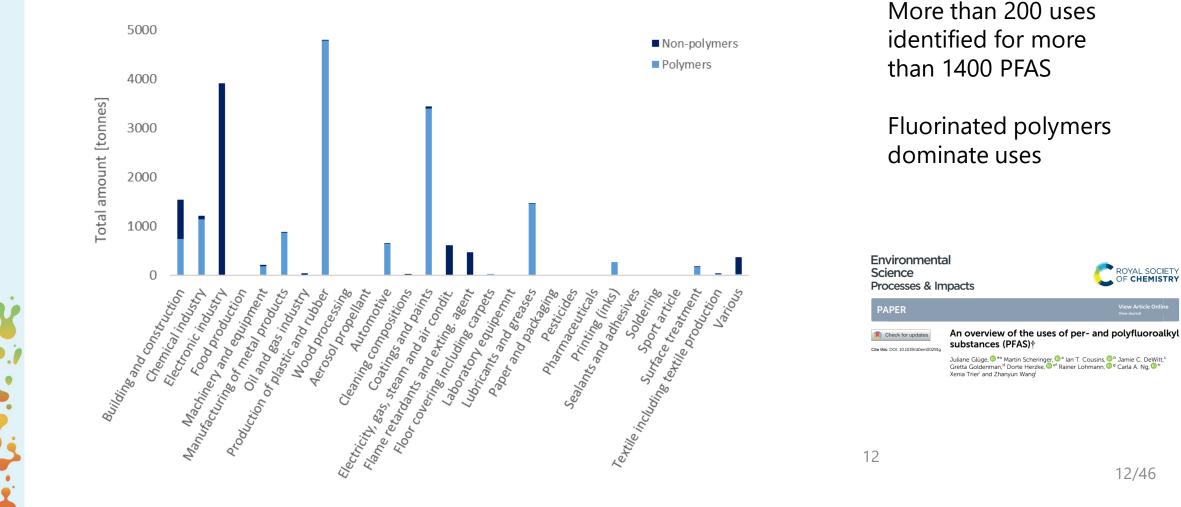
#### **REVIEW SUMMARY**

Per- and polyfluoroalkyl substances in the environment

Marina G. Evich†, Mary J. B. Davis†, James P. McCord†, Brad Acrey, Jill A. Awkerman, Detlef R. U. Knappe, Andrew B. Lindstrom, Thomas F. Speth, Caroline Tebes-Stevens, Mark J. Strynar, Zhanyun Wang, Eric J. Weber, W. Matthew Henderson\*, John W. Washington\*

# Uses of PFAS

### (based on information in Substances in Preparations In Nordic countries (SPIN) database)



# **Madrid Statement**

- Published in 2015
- Signed by 250 scientists from 38 countries
- Production and use of PFAS should be limited
- But are all PFAS problematic?
  - they have diverse properties, right?

### Perspectives | Brief Communication

#### The Madrid Statement on Poly- and Perfluoroalkyl Substances (PFASs)

http://dx.doi.org/10.1289/ehp.1509934

Arlene Blum,<sup>1,2</sup> Simona A. Balan,<sup>2</sup> Martin Scheringer,<sup>3,4</sup> Xenia Trier,<sup>5</sup> Gretta Goldenman,<sup>6</sup> Ian T. Cousins,<sup>7</sup> Miriam Diamond,<sup>8</sup> Tony Fletcher,<sup>9</sup> Christopher Higgins,<sup>10</sup> Avery E. Lindeman,<sup>2</sup> Graham Peaslee,<sup>11</sup> Pim de Voogt,<sup>12</sup> Zhanyun Wang,<sup>4</sup> and Roland Weber<sup>13</sup>

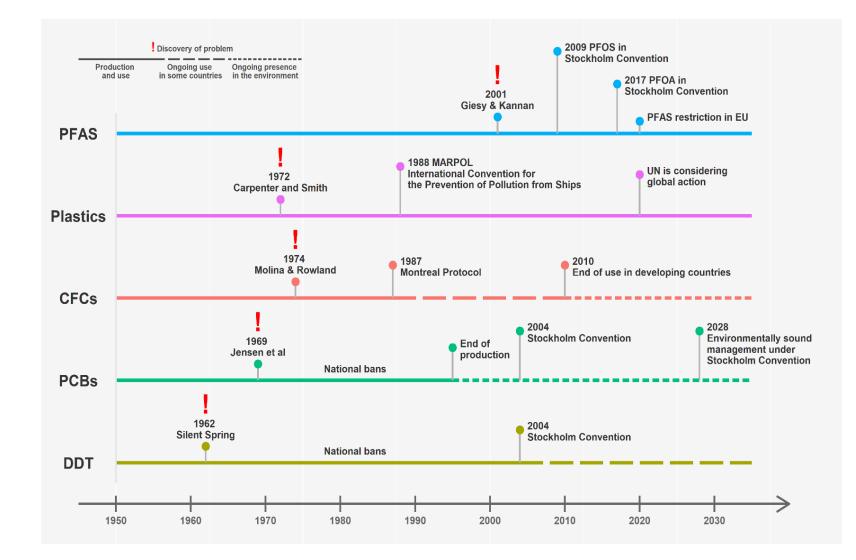
## **Regulating all PFAS?**

- Authorities of Denmark, Germany, Netherlands, Norway and Sweden published a REACH restriction proposal for all PFAS meeting the OECD definition
  - Published in January 2023 and was under public consultation until 25 September 2023
  - Broad restriction based on known high persistence of all PFAS and unknown bioaccumulation potential and toxicity of most PFAS
  - 5700 comments (mostly from industry) being processed
- Separate parallel restriction proposal on PFAS in firefighting foams
  - Nearly finalized
  - 10 year phase out of PFAS in firefighting foams
- California: ban PFAS in all consumer products

## Are all PFAS of concern?

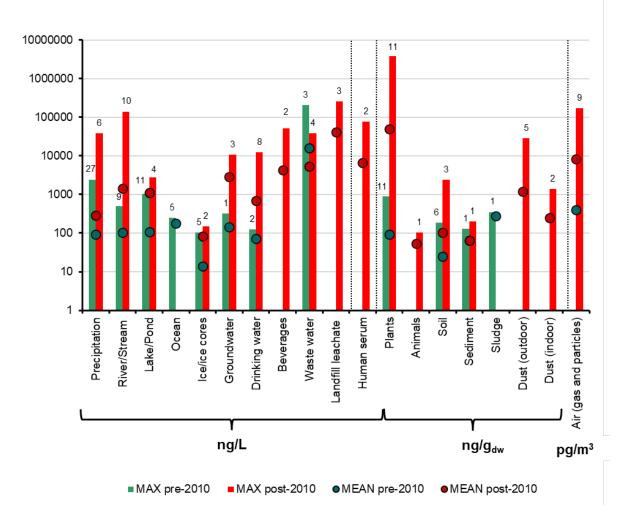
- All PFAS are very persistent (vP) (EU REACH)
  - they are either non-degradable or transform into stable terminal transformation products
  - they are all extremely P with no environmental degradation observed, due to C-F bonds
- Continual release of high P chemicals results in increasing levels and increasing probabilities of known and unknown effects.
   Exposure poorly reversible
- They are synthetic (few natural sources), thus "novel entities"

## **Problems with high persistence**

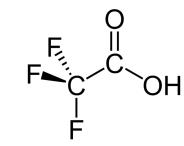


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# TFA the "poster child" for the problem with high persistence



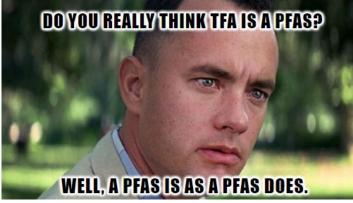
TFA concentrations in different media before 2010 (in green) and after 2010 (in red).



TFA – trifluoroacetic acid

## **Sources of TFA?**

- Many "precursors"
  - refrigerants, pesticides, pharmaceuticals and other industrial chemicals
  - HFO-1234-yf, used as refrigerant, is a notable precursor
  - emissions of HFO-1234-yf projected to increase by factor of 7 between 2020 and 2050
- Destructive treatments of PFAS can also be a source of TFA (e.g. oxidation, incineration, electrolysis).
- Emissions hotspots include AFFF-contaminated sites, landfills and fluorochemical production facilities.

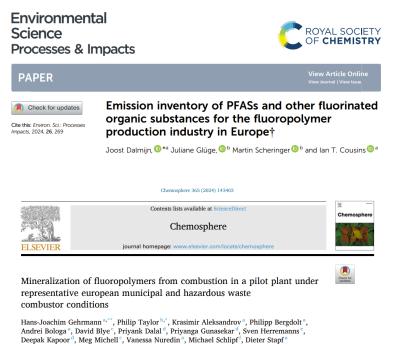


## Are fluoropolymers a problem?

## • Fluoropolymers (e.g. PTFE)

 high molecular weight, stable, inert, insoluble, involatile, immobile, do not cross biological membranes, low leachables

- Emissions during lifecycle
  - emissions of known and unknown PFAS during manufacturing
  - emissions of low molecular weight PFAS during incineration?



## **PFAS** being regulated in Europe

#### 24 France 24

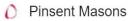
#### 'Forever chemicals': French MPs approve PFAS product ban

French MPs on Thursday approved the first reading of a bill aimed at restricting the manufacture and sale of non-essential products...

#### JS Just Style

#### Denmark to ban PFAS in clothing, shoes from July 2026

As part of a new PFAS action plan Denmark's government has proposed to introduce a national ban on PFAS in clothing and shoes.



#### New UK PFAS restrictions closer as debate over alternatives intensifies

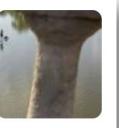
Manufacturers could face restrictions on using per- and polyfluoroalkyl substances (PFAS) - so-called 'forever chemicals' - to make cleaning...











### **ECHA**

ANNEX	xν	RESTRICTION	REPORT

**PROPOSAL FOR A RESTRICTION** 

SUBSTANCE NAME(S): Per- and polyfluoroalkyl substances (PFASs) IUPAC NAME(S): n.a.

EC NUMBER(S): n.a.

CAS NUMBER(S): n.a.

#### CONTACT DETAILS OF THE DOSSIER SUBMITTERS:

#### BAuA Federal Institute for Occupational Safety and Health **Division 5 - Federal Office for Chemicals** Friedrich-Henkel-Weg 1-25 D-44149 Dortmund, Germany

Bureau REACH, National Institute for Public Health and the Environment (RIVM) Antonie van Leeuwenhoeklaan 9 3721 MA Bilthoven, The Netherlands

Swedish Chemicals Agency (KEMI) PO Box 2. SE-172 13 Sundbyberg, Sweden

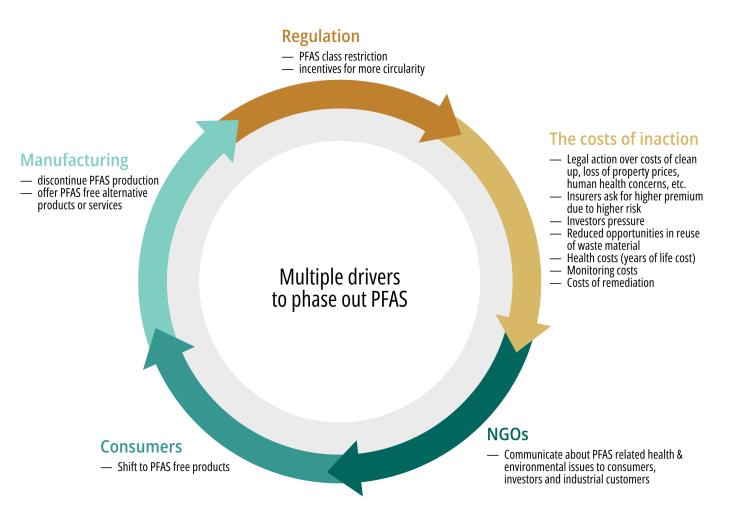
Norwegian Environment Agency P.O. Box 5672 Torgarden N-7485 Trondheim, Norway

The Danish Environmental Protection Agency Tolderlundsvej 5 5000 Odense C, Denmark

#### **VERSION NUMBER: 2**

DATE: 22.03.2023

# Drivers to phase out PFAS and find alternatives



## Essentiality

- Impractical to ban all uses of all PFAS in one step
- Can "essentiality" guide substitution?
- Montreal Protocol on Substances that Deplete the Ozone Layer
  - Introduced concept of essential use
  - Adapt the Montreal Protocol's criteria for essentiality
  - "necessary for health and safety or critical to the functioning of society"

# **Defining essentiality**

Category

#### Definition

1	
"Non-essential"	

Uses that are not essential for health and safety, and the functioning of society. The use of substances is driven primarily by market opportunity.

2 "Substitutable" Uses that have come to be regarded as essential by society because they perform important functions, but where alternatives to the substances have now been developed that have equivalent functionality and adequate performance, which makes those uses of the substances no longer essential.

- 3 "Essential"
- Uses considered essential by society because al" they are necessary for health or safety or other highly important purposes *and* for which alternatives are not yet established.\*

#### Environmental Science Processes & Impacts

#### **CRITICAL REVIEW**

() Check for updates

Cite this: Environ. Sci.: Processes Impacts, 2019, 21, 1803

### The concept of essential use for determining when uses of PFASs can be phased out

lan T. Cousins, <sup>(ij)</sup>†<sup>\*a</sup> Gretta Goldenman,<sup>b</sup> Dorte Herzke,<sup>c</sup> Rainer Lohmann, <sup>(jj)</sup><sup>d</sup> Mark Miller,<sup>e</sup> Carla A. Ng, <sup>(ij)</sup> Sharyle Patton,<sup>g</sup> Martin Scheringer, <sup>(ij)</sup><sup>h</sup> Xenia Trier,<sup>i</sup> Lena Vierke,<sup>j</sup> Zhanyun Wang <sup>(ij)</sup><sup>k</sup> and Jamie C. DeWitt<sup>1</sup>

Applied the concept to multiple cases studies for use cases of PFAS

\*This essentiality should not be considered permanent; rather, a constant pressure is needed to search for alternatives in order to move these uses into Category 2 above.

OYAL SOCIETY

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## It's all about function

- What is the function of the chemical of concern?
- Different aspects of function ("functional substitution")
  - Chemical function, end-use function and service function (from "functional substitution")
- Function of PFAS in aqueous-film forming foams (AFFF)
  - chemical function = powerful surfactant, inert (persistent!)
  - end-use function = to facilitate the spreading of an aqueous film that rapidly spreads over the fuel and flames
  - service function = to rapidly extinguish a fuel fire



Policy Analysis pubs.acs.org/est

Advancing Safer Alternatives Through Functional Substitution Joel A. Tickner,<sup>\*,†</sup> Jessica N. Schifano,<sup>‡</sup> Ann Blake,<sup>§</sup> Catherine Rudisill,<sup>||</sup> and Martin J. Mulvihill<sup>⊥</sup>

## **Opportunities for substitution**

#### Table 1. Functional Substitution for Chemicals in Products, Chemicals in Processes

Functional	Chemical in Product	Chemical in Process		
Substitution Level	Bisphenol-a in Thermal Paper	Methylene Chloride in Degreasing Metal Parts		
Chemical Function (Chemical Change)	Is there a functionally equivalent chemical substitute (i.e., chemical developer)? Result: Drop-in chemical	Is there a functionally equivalent chemical substitute (i.e., chlorinated solvent degreaser)? Result: Drop-in chemical		
	replacement	replacement		
End Use Function (Material, Product, Process Change)	Is there another means to achieve the function of the chemical in the product (i.e., creation of printed image)?	Is there another means to achieve the function of the process (i.e., degreasing)?		
	Result: Redesign of thermal paper, material changes	<b>Result: Redesign of the process</b> (e.g., ultrasonic, aqueous)		
<b>Function As Service</b> (System Change)	Are cash register receipts necessary? Are there alternatives that could achieve the same purpose (i.e. providing a record of sale to a consumer)?	Is degreasing metal parts necessary? Are there other alternatives that could achieve the same purpose (i.e., providing metal parts free of contaminants for other end uses)?		
	Result: Alternative printing systems (e.g., electronic receipts)	Result: Alternative metal cutting methods		





Advancing Safer Alternatives Through Functional Substitution

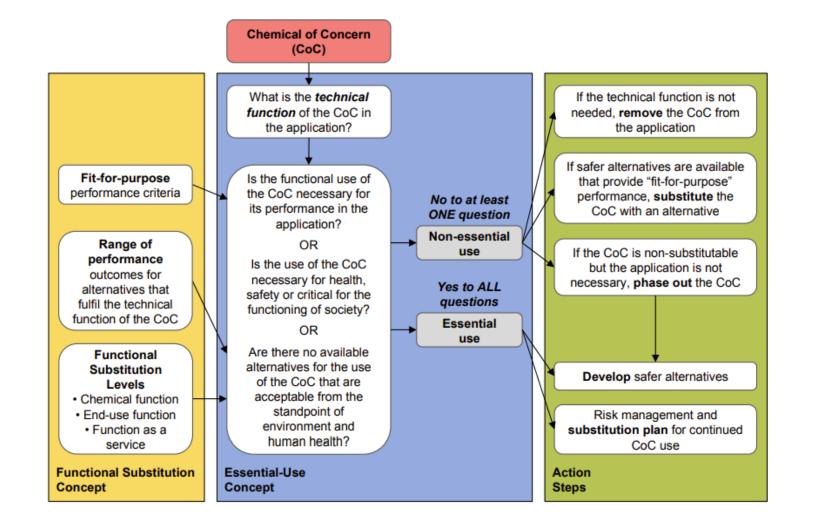
Joel A. Tickner,\*<sup>,†</sup> Jessica N. Schifano,<sup>‡</sup> Ann Blake,<sup>§</sup> Catherine Rudisill,<sup>∥</sup> and Martin J. Mulvihill<sup>⊥</sup>

## Aspects to consider

- Aspect 1: Is the functional use necessary for its performance in the application?
- Aspect 2: Is the use necessary for health and safety and critical for the functioning of society?
- Aspect 3: Are there no available alternatives for this particular use that are acceptable from the viewpoint of health safety, etc.?
- If all aspects true then **essential use**
- If one false then **non-essential use**

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# Combining functional substitution and the essential-use concept





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Viewpoint

#### Combined Application of the Essential-Use and Functional Substitution Concepts: Accelerating Safer Alternatives

Monika A. Roy, Ian Cousins, Elizabeth Harriman, Martin Scheringer, Joel A. Tickner,  $\!\!\!^*$  and Zhanyun Wang

## **Currently Unavoidable Uses (CUUs)**

- Maine and Minnesota set March 1 2024 for comments and submissions related to identification of currently unavoidable uses (CUUs) of PFAS in products.
  - Maine's PFAS in Products law, 38 M.R.S. § 1614
  - PFAS in Products law, Minn. Stat. § 116.943
- Ultimately, the goal of the CUU process will be securing exemption from upcoming restrictions on products which contain PFAS.
  - Public participation in these processes
  - Maine and Minnesota are cooperating in the proceses

# Three example case studies for uses of PFAS

## Ski waxes

- Fluorinated waxes are favored by competitive skiers
- Waxes contain fluorinated alkanes, but also perfluoroalkyl carboxylates, including PFOA
- Don't need to ski e.g. 5% faster. Alternatives "fit for purpose"



FIS ban on fluorinated ski waxes in place since 2020/21 season

Delay due to lack of testing methods

Skis tested for F-content in less than one minute using FT-IR spectroscopy

# AFFFs (aqueous film-forming foams)

- Extinguishing class B fuel fires
- Fluorine-free class B foams (3F) available since early 2000s
  - meet the standard firefighting performance certifications
  - use different end-use function than AFFFs
  - many commercial airports have phased out AFFFs and use 3F
- Still some debate if 3F suitable for certain scenarios
  - Are they "fit for purpose"?



## **Repellency in textiles**

Personal and

occupationa

protection

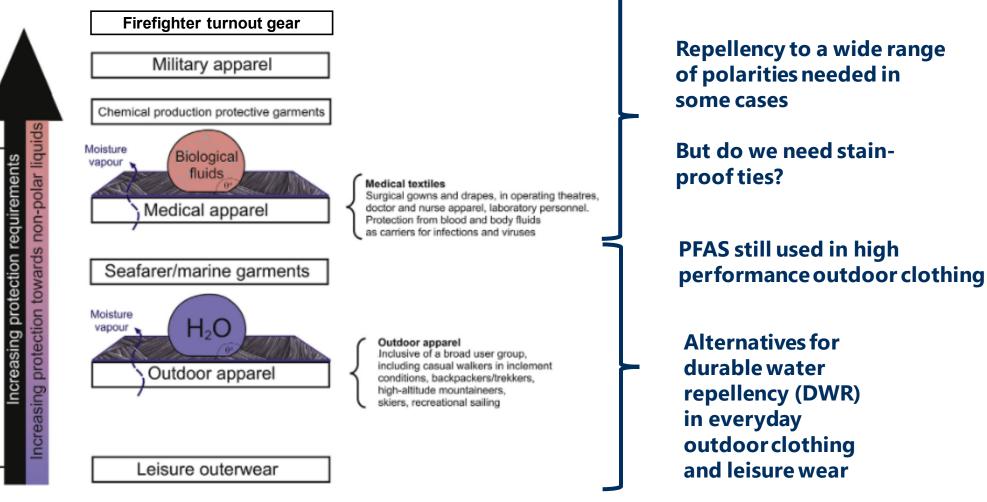
Aesthetic and

easy care



Highly fluorinated chemicals in functional textiles can be replaced by re-evaluating liquid repellency and end-user requirements

Steffen Schellenberger $^{a,\,1},$  Philippa J. Hill $^{b,\,1},$  Oscar Levenstam $^c,$  Philip Gillgard $^d,$ Ian T. Cousins $^{a,\,*},$  Mark Taylor $^b,$  Richard S. Blackburn  $^{b,\,**}$ 

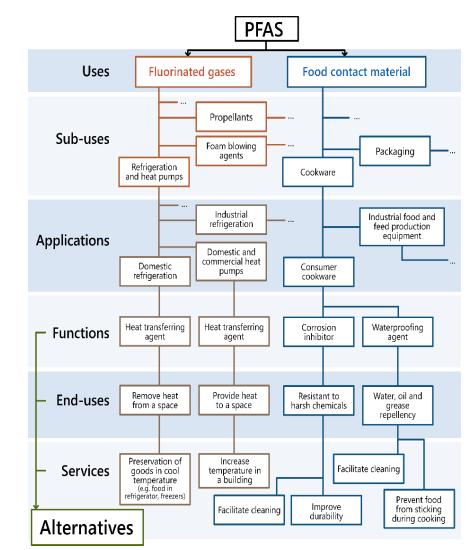


# Challenges for applying essentiality

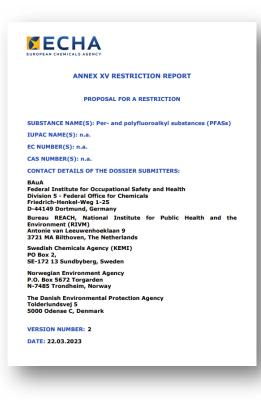
- Substitutable uses
  - avoid regrettable substitutions: alternatives assessment
- How to decide when uses are "essential" or "nonessential"?
  - in many use cases there are alternatives
  - but when there are no alternatives who decides what is necessary for health and safety and critical for the functioning of society?
  - role of multiple stakeholders, social science?

## **PFAS** alternatives database

- Database divided per use category of PFAS
- For each use category:
  - List of applications of PFAS along with the chemical function, end use function and service function
  - List of PFAS substances identified as being used for each application
  - List of identified potential alternatives to PFAS for each application



# Important sources of information







### 36/46

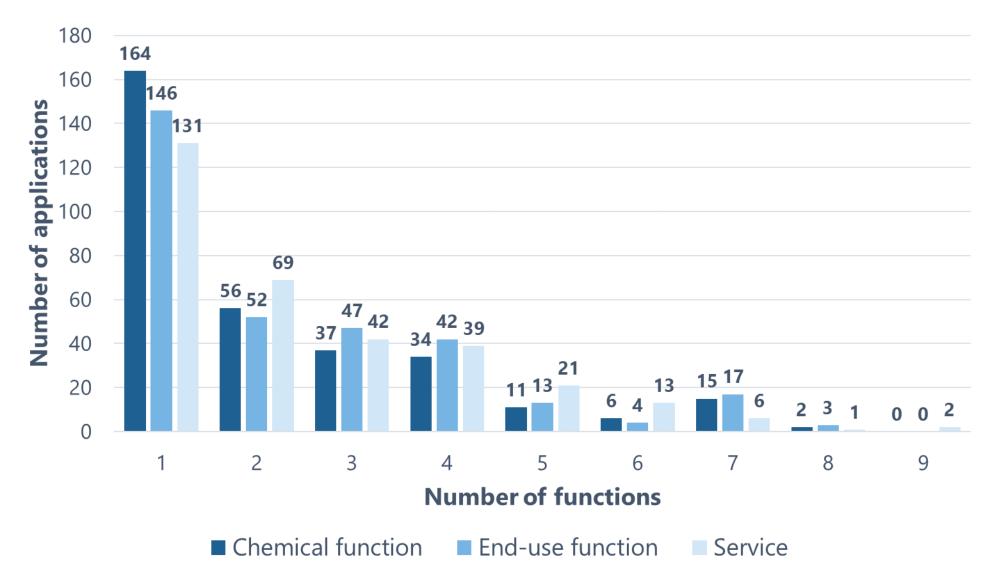
# Uses and functions of PFAS

- 18 use categories of PFAS included in the database
- Waterproofing agent, heat stabilizer and corrosion inhibitor are the most common technical functions
- Mainly for enhancing durability

Use categories	Sub-uses	Applications	Technical functions		Services
Active pharmaceutical ingredients	2	14	1	24	21
Biocides	1	4	1	4	4
Building and construction products	9	17	14	18	30
Consumer mixtures	7	15	7	15	12
Cosmetic products	5	32	9	12	6
Electronics and semiconductors sector	3	29	17	22	37
Energy sector	9	19	17	19	24
Firefighting foams	1	5	1	1	3
Fluorinated gases	7	29	8	14	27
Food contact materials	2	4	4	4	9
Industrial production	8	28	12	18	11
Lubricants	3	42	11	13	19
Medical products	6	21	14	18	29
Metal plating and metal products manufacture	2	4	8	10	14
Petroleum and mining	2	9	10	13	15
Plant Protection Products	1	6	3	7	5
Textile, upholestry, leather, apparel, and carpets	7	20	11	15	21
Transport sector	10	27	16	24	45
Grand Total	85	325	39	131	201



## Number of functions provided by PFAS



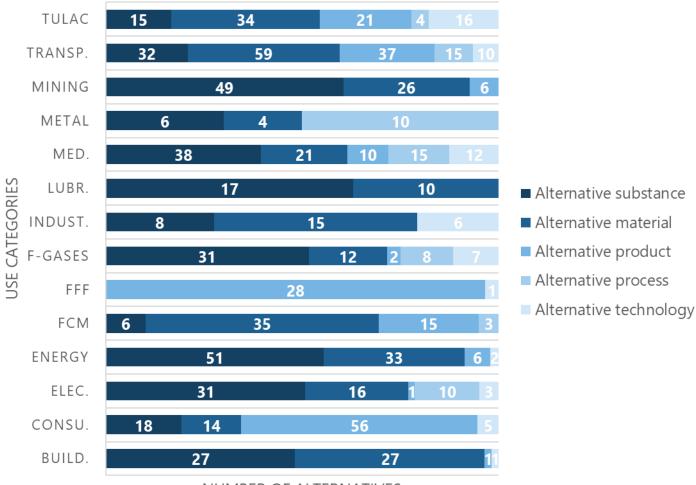
## **Information on alternatives**

- **Alternative identifiers:** name and CAS number (if relevant)
- Alternative type: substance; material; product; process; technology
- General chemistry: Organic synthetic; silicone-based; natural-based; inorganic
- PBT assessment and CLP classification (if available)
- Flagged: whether an alternative is listed in the Substitution Support Portal (<u>https://www.subsportplus.eu/subsportplus/EN/Substances/Database-of-</u> restricted-and-priority-substances/restricted-priority-substances\_node.html)
- **Change in functionality and market uptake:** Based on information available in the PFAS restriction



## Number and types of alternatives to PFAS

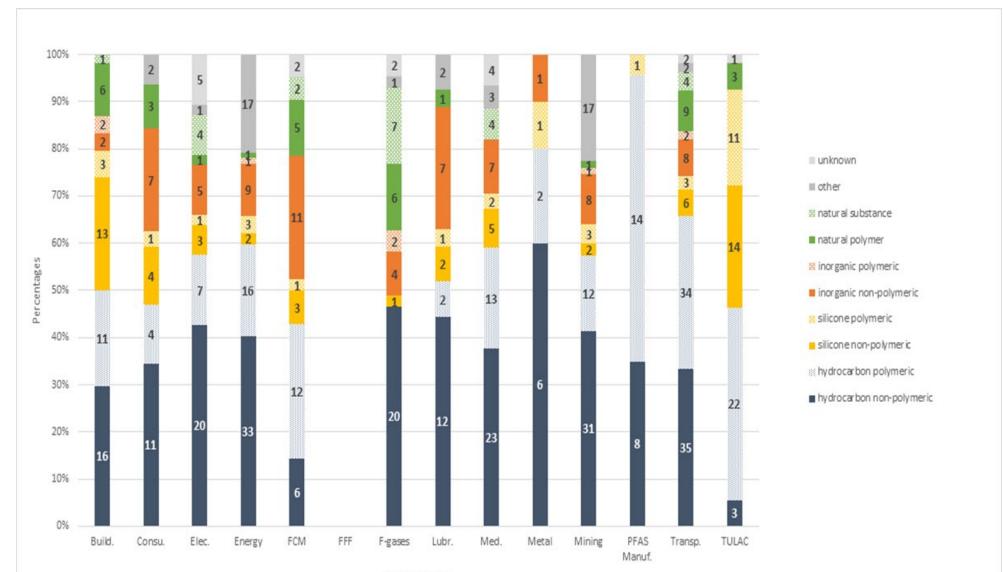
- Database includes 532 different potential alternatives across 14 use categories
  - Left out active pharmaceutical ingredients, cosmetic products, plant protection products and biocides
- No alternative found for 83 applications (mainly for industrial production)



NUMBER OF ALTERNATIVES

## **Alternative chemistries**

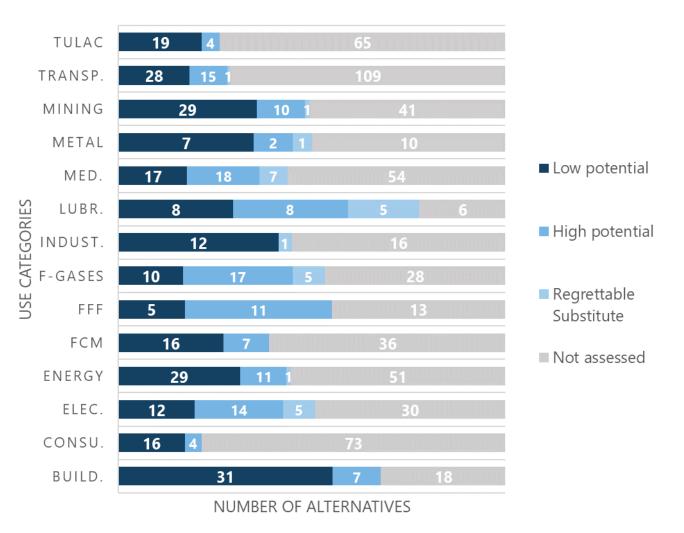
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Use categories

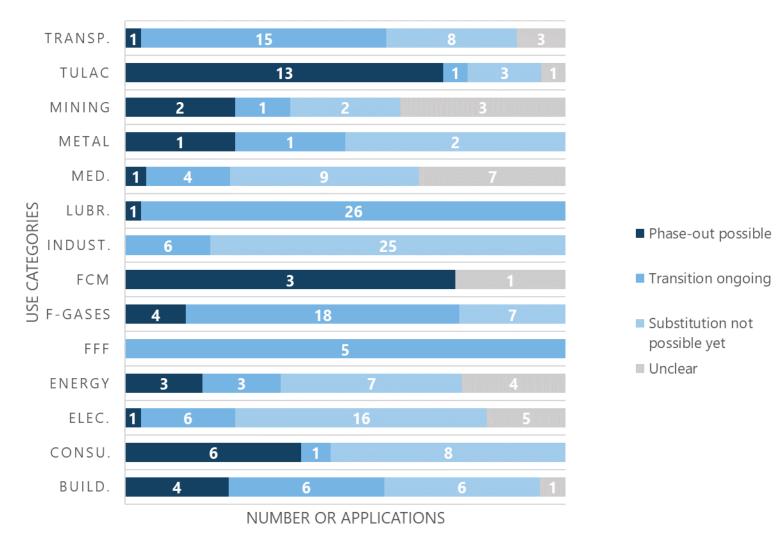
## Safety of alternatives to PFAS

- Evaluation based on the Substitution Support Portal (https://www.subsportplus.eu/su bsportplus/EN/Home/Home\_no de.html)
- Possible only for alternatives identified by a CAS number (i.e. 36%)



## Are the alternatives suitable and available?

 Overall, no suitable alternatives available for ~25% of PFAS applications



## **Biggest challenges to universal phase-out**

- F-gases
- Industrial uses of fluoropolymers
- Green energy transition
- Medical applications



Home | News articles | Importance of fluoropolymers for the clean energy transition and the EU's net zero industry

## Importance of fluoropolymers for the clean energy transition and the EU's net zero industry







Fluoropolymers - Irreplaceable uses - EU PFAS restriction - Knowledge center FAQ - News About Contact



#### The importance of fluoropolymers for the sector

Europe is at the forefront of the development and the use of renewable energy and has one of the most ambitious goals for a sustainable future in this respect. Houropalymers are an indispensable driver of the European Green Deal, as they have a key role to play in the transition to an integrated and decarbonized energy system. With their unique combination of properties, lluoropolymers are crucial components in hydrogen

# **Conclusions/observations**

- It's encouraging that so many alternatives to uses of PFAS already available and transitions have occurred or are happening
  - Hard work for product manufacturers and retailers
- Challenges remain but shift happens!
  - Unique properties of PFAS offering multiple functions
  - Some applications with no available alternatives currently available
  - Ensuring no regrettable substitutions (alternatives assessment)
  - Resistance from some industries
- Green energy transition being used as a "free pass" for derogations
- Strong PFAS regulation, and other drivers, encourage innovation

## For more information



## Email: <u>romain.figuiere@aces.su.se</u>

## Access to the alternatives

database



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